

SYNTHESIS OF 3 SUBSTITUTED ISOQUINOLIN-1-YL-2-(CYCLOALK-2-ENYLIDENE) HYDRAZINES AND THEIR ANTIMICROBIAL PROPERTIES

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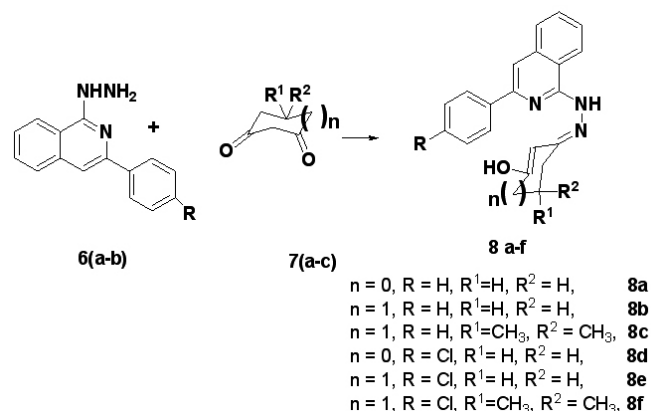
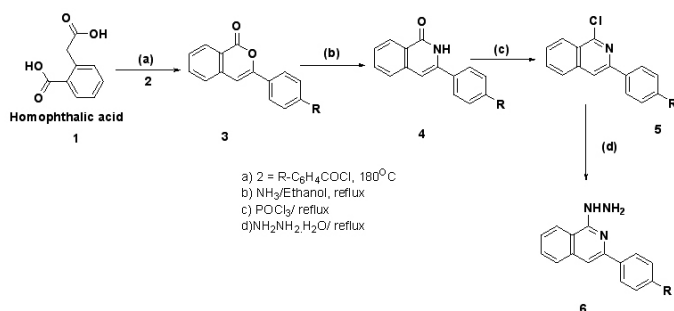
ABSTRACT

New hydrazone derivatives were synthesized via reaction between 1, 3 cyclic diketones and hydrazinoisoquinoline derivatives. The reaction proceeded smoothly in ethanol under reflux temperature and furnished products in excellent yields (76–87%). The products have been purified and fully characterized by spectroscopy techniques. The compounds **8a-c** showed good bacterial inhibition against *Bacillus cerus* and **8d-f** showed good antifungal activity against *Candida albicans*.

Key words: 1, 3-cyclic diketones, hydrazinoisoquinoline derivatives.

INTRODUCTION

Azomethine group bearing compounds are usually synthesized from the condensation of primary amines and active carbonyl groups. They are important class of compounds in medicinal and pharmaceutical field. They show biological applications including antibacterial, antifungal and antitumor activity¹⁻⁴. A variety of pharmacological effects are associated with isoquinoline derivatives including sedative, hypotensive, neuromuscular blocking and CNS activities⁵. In recent years, there has been significant interest in the synthesis of these compounds and many approaches have been reported^{6,7}. Although Schiff bases and azomethine derivatives are among the most thoroughly studied compounds, we were surprised that there has been no report of the isoquinolinyl-substituted azomethine derivatives. As part of a program to synthesize new heterocyclic compounds as potential pharmaceuticals⁸⁻¹⁵, we have investigated this reaction. We herein report the results of this study. The titled compounds were synthesized from homophthalic acid by five step synthesis. Homophthalic acid (**1**), on refluxing with acid chloride (**2**) in the absence of solvent yielded 3-substituted isocoumarin⁸ (**3**). Compound **3** on reaction with ammonia in the presence of ethanol yielded 3-substituted isoquinolinone (**4**). Compound **4** in presence of POCl₃ yielded 1-Chloro-3-substituted isoquinoline¹⁶ (**5**), which on reaction with hydrazine hydrate yielded 1-hydrazino-3-substituted isoquinoline¹⁷⁻¹⁹ (**6**). The titled compounds (**8**) were prepared by treating (**6**) with appropriate 1, 3 diketones (**7**). The purity of synthesized compounds was monitored by thin layer chromatography (TLC) and LCMS elemental analyses and structures were identified by spectral data.



Scheme: 1 Synthesis of Isoquinolin-1-yl-2-(cycloalk-2-enylidene) hydrazines

EXPERIMENTAL SECTION

Materials and Methods

Chemicals were purchased from Aldrich Chemical Co. and used as such without further purification. TLC was performed on silica plates with visualization by UV-light. Melting points were taken in open capillary tubes and corrected with reference to benzoic acid. IR spectra in KBr pellets were recorded on Nucon Infrared spectrophotometer. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded on a Bruker 400 MHz spectrometer in CDCl₃ or DMSO (with TMS for ¹H NMR and DMSO for ¹³C NMR as internal references). LC-MS analyses were performed with LCMS-Agilent- 1100 series Ion Trap.

Synthesis of Isoquinolin-1-yl-2-(cycloalk-2-enylidene) hydrazines (8a-f)

The hydrazone, (**6a-b**) and 1, 3-diketones, (**7a-c**) were taken in ethanol (1:1 ratio) and refluxed under nitrogen atmosphere for overnight. Then reaction mass was quenched with water, extracted by AcOEt (20 mL X 3), washed, dried, concentrated and purified by column chromatography using silica gel to get the Schiff bases, (**8a-f**). The products obtained were characterized by IR, LCMS, ¹H-NMR and ¹³C-NMR techniques. The reaction of hydrazone **6 a-b** with various diketones, **7a-c** was tabulated (Table 1).

The spectral data of compounds, **8a-8f** is given below.

(1Z)-2-(cyclo-3-hydroxy-pent-2-enylidene)-1-(3-phenylisoquinolin-1-yl)hydrazine (8a) Dark brown solid, IR cm⁻¹ 3438 (-OH), 3326 (-NH), 1714 (C=C), 1649 (C=N); ¹H NMR (400 MHz, DMSO-d₆) 9.73 (s, 1H, -OH), 9.51 (s, 1H), 8.31 (s, 1H), 8.25-8.23 (d, J=8.0 Hz, 2H), 8.16-8.14 (d, J=8.0 Hz, 1H), 7.90-7.88 (d, J=8.2 Hz, 1H), 7.78 (s, 1H, -NH), 7.72-7.68 (t, J=7.5 Hz, 1H), 7.57-7.53 (t, J=4.0 Hz, 2H), 7.47-7.44 (m, 1H), 7.38-7.34 (t, J=7.2 Hz, 1H), 4.84 (s, 1H, C=CH), 2.74-2.66 (m, 2H, C-CH₂), 2.28 (m, 2H, C-CH₂);

¹³C NMR (100 MHz, DMSO-d₆) 202.35 (C=COH), 154.01 (C=C-NH), 147.84 (C-C=N), 139.46, 138.20, 133.46, 130.90, 128.97, 2 X (128.79), 2 X (127.83), 126.71, 123.17, 116.51, 108.67 (aromatic carbons), 99.35, 79.63, 34.15, 25.76 (alicyclic carbons); LCMS: m/e 316.12; Mol. Formula C₂₀H₁₇N₃O, Mol. Wt.: 315.37.

Table 1- Synthesis of Isoquinolin-1-yl-2-(cycloalk-2-enylidene)hydrazines from 1, 3 cyclic diketones and hydrazinoisoquinoline derivatives.

Entry	Compounds ^a	R	n	R ¹	R ²	mp ^o C	^b Yield %
1	8a	H	0	H	H	201-203 ^o C	80
2	8b	H	1	H	H	191-193 ^o C	82
3	8c	H	1	CH ₃	CH ₃	205-207 ^o C	85
4	8d	Cl	0	H	H	235-237 ^o C	76
5	8e	Cl	1	H	H	193-195 ^o C	83
6	8f	Cl	1	CH ₃	CH ₃	215-217 ^o C	87

^aAll products were identified by ¹H, ¹³C NMR, ^bisolated yields.

(17Z)-2-(cyclo-3-hydroxy-hex-2-enylidene)-1-(3-phenylisoquinolin-1-yl)hydrazine (8b) Brown solid, IR cm⁻¹ 3324 (-OH), 3212 (-NH), 1613 (C=N), 1583 (C=C); ¹H NMR (400 MHz, DMSO-d₆) 9.55 (s, 1H, -OH), 9.01 (s, 1H), 8.28-8.26 (d, J=8.3 Hz, 1H), 8.17-8.15 (d, J=8.2 Hz, 2H), 7.89-7.87 (d, J=8.0 Hz, 1H), 7.77 (s, 1H, -NH), 7.72-7.68 (t, J=7.5 Hz, 1H), 7.57-7.53 (t, J=4.0 Hz, 1H), 7.48-7.44 (m, 2H), 7.38-7.34 (t, J=7.2 Hz, 1H), 5.01 (s, 1H, C=CH), 2.55 (m, 2H, C-CH₂), 2.14-2.11 (m, 2H, C-CH₂), 1.94-1.93 (d, 2H, C-CH₂); ¹³C NMR (100 MHz, DMSO-d₆) 195.35(C=COH), 164.76, 154.01, 147.86(C-C=N), 139.53, 138.25, 130.86, 128.96, 2 X (128.79), 2 X (127.87), 126.69, 123.09, 116.49, 108.45 (aromatic carbons), 96.41, 79.63, 37.22, 26.63, 22.34 (alicyclic carbons); LCMS: m/e 330.1; Mol. Formula C₂₁H₁₉N₃O, Mol. Wt.: 329.4.

(17Z)-2-(5,5-dimethylcyclo-3-hydroxy-hex-2-enylidene)-1-(3-phenylisoquinolin-1-yl)hydrazine (8c) Pale yellow, IR cm⁻¹ 3217(-OH), 3032(-NH), 1625(C=N), 1593(C=C); ¹H NMR (400 MHz, DMSO-d₆) 9.55 (s, 1H, -OH), 8.99 (s, 1H), 8.29-8.27 (d, J=8.3 Hz, 1H), 8.19-8.17 (t, J=4.2 Hz, 2H), 7.89-7.87 (d, J=8.0 Hz, 1H), 7.78 (s, 1H, -NH), 7.71-7.68 (t, J=7.5 Hz, 1H), 7.56-7.52 (m, 1H), 7.44-7.41 (m, 2H), 7.36-7.33 (t, J=7.2 Hz, 1H), 4.98 (s, 1H, C=CH), 2.42 (s, 2H, C-CH₂), 2.01 (s, 2H, C-CH₂), 1.07 (s, 6H, -CH₃); ¹³C NMR (100 MHz, DMSO-d₆) 194.91(C=COH), 164.76(C=C-NH), 154.07(CH₂-C=N), 147.83(C=N), 133.42, 139.39, 138.22, 130.84, 128.83, 2 X (128.78), 2 X (127.86), 126.71, 123.07, 116.47, 108.39 (aromatic carbons), 94.83, 79.67, 51.08, 33.10 (alicyclic carbons), 2 X 28.54 (-CH₃); LCMS: m/e 358.1; Mol. Formula C₂₃H₂₃N₃O, Mol. Wt.: 357.45.

(18Z)-1-(3-(4-chlorophenyl)isoquinolin-1-yl)-2-(3-hydroxy-cyclohex-2-enylidene)hydrazine (8d) Pale yellow, IR cm⁻¹ 3428 (-OH), 3349 (-NH), 1705 (C=N), 1651(C=C); ¹H NMR (400 MHz, DMSO-d₆) 9.77 (s, 1H, -OH), 9.52 (s, 1H), 8.25-8.23 (d, J=8.3 Hz, 1H), 8.18-8.16 (d, J=8.5 Hz, 2H), 7.89-7.87 (d, J=8.1 Hz, 1H), 7.81 (s, 1H, -NH), 7.73-7.69 (t, J=7.5 Hz, 1H), 7.58-7.51 (m, 3H), 4.83 (s, 1H, =CH), 2.75 (s, 2H, -CH₂), 2.29 (s, 2H, -CH₂); ¹³C NMR (100 MHz, DMSO-d₆) 202.44 (C=COH), 178.57(C=C-NH), 154.05(CH₂-C=N), 146.59(-C=N), 138.36, 138.12, 133.46, 131.00, 128.99, 2 X (128.37), 2 X (127.90), 126.95, 123.19, 116.63, 108.85 (aromatic carbons), 99.36, 34.15, 25.75(alicyclic carbons); LCMS: m/e 350.0; Mol. Formula C₂₀H₁₆ClN₃O, Mol. Wt.: 349.81.

(18Z)-1-(3-(4-chlorophenyl)isoquinolin-1-yl)-2-(3-hydroxy-cyclohex-2-enylidene)hydrazine (8e) Pale yellow, IR cm⁻¹ 3331(-OH), 3230 (-NH), 1685(C=N), 1584 (C=C); ¹H NMR (400 MHz, DMSO-d₆) 9.58 (s, 1H, -OH), 9.00 (s, 1H), 8.28-8.26 (d, J=8.3 Hz, 1H), 8.18-8.16 (d, J=8.6 Hz, 2H), 7.89-7.87 (d, J=8.1 Hz, 1H), 7.80 (s, 1H, -NH), 7.72-7.69 (t, J=7.5 Hz, 1H), 7.58-7.51 (m, 3H), 4.99 (s, 1H, C=CH), 2.55 (t, 2H, -CH₂), 2.14-2.11 (m, 2H, -CH₂), 1.92 (t, 2H, -CH₂); ¹³C NMR (100 MHz, DMSO-d₆) 195.39(C=COH), 146.59(C=C-NH), 164.76, 154.01, 138.41, 138.13, 133.44, 130.98, 128.97, 2 X (128.35), 2 X (127.92), 126.93, 123.10, 116.58, 108.65 (aromatic carbons), 96.41, 79.63, 37.22, 22.34 (alicyclic carbons); LCMS: m/e 364.0; Mol. Formula C₂₁H₁₈ClN₃O, Mol. Wt.: 363.84.

(18Z)-1-(3-(4-chlorophenyl)isoquinolin-1-yl)-2-(3-hydroxy-5,5-dimethylcyclohex-2-enylidene)hydrazine (8f) Pale yellow, IR cm⁻¹ 3210 (-OH), 3027 (-NH str.), 1626 (C=N), 1594 (C=C); ¹H NMR (400 MHz, DMSO-d₆) 9.59 (s, 1H, -OH), 8.99 (s, 1H), 8.31-8.27 (t, J=7.3 Hz, 1H), 8.22-8.18 (m, 2H), 7.89-7.87 (d, J=8.0 Hz, 1H), 7.81 (s, 1H, -NH), 7.73-7.69 (m, 1H), 7.58-7.54 (m, 1H), 7.51-7.48 (m, 2H), 4.96 (s, 1H, =CH), 2.42 (s, 2H, -CH₂), 2.01 (s, 2H, -CH₂), 1.07 (s, 6H, -CH₃); ¹³C NMR (100 MHz, DMSO-d₆) 194.96(C=COH), 164.71(C=C-NH), 154.13(CH₂-C=N), 146.54(-C=N), 138.28, 138.13, 133.44, 130.98, 128.85, 2 X (128.38), 2 X (127.93), 126.95, 123.10, 116.57, 108.62 (aromatic carbons), 94.82, 79.63, 51.07, 33.13 (alicyclic carbons), 28.50(-CH₃), 28.39(-CH₃); LCMS: m/e 392.0; Mol. Formula C₂₃H₂₂ClN₃O, Mol. Wt.: 391.89.

Antimicrobial properties

Cultures of bacteria were grown on nutrient broth (HiMedia) at 37°C for 12 – 14 hr and that of fungus on Sabouraud dextrose broth (HiMedia) at 28°C for 48 hr and were maintained on respective agar slants at 4°C. The compounds **8a-f** were screened for their antibacterial activities against *Escherichia coli*, *Salmonella typhi*, *Proteus mirabilis*, *Bacillus cerus*, *Staphylococcus aureus*, and antifungal activities against *Candida albicans*, *Aspergillus flavus*, *Aspergillus niger* by agar well technique²⁰. Standard antibacterial eftazidime, Chloramphenicol and antifungal Nalidixic acid were also tested under similar conditions for comparison. The compounds **8a-f** of 4 mg/ml concentration was used as stock solution, from this 100 µl was loaded to each well. The antimicrobial properties were duplicated and the averages were taken.

RESULTS AND DISCUSSION

The 1-hydrazino-3-substituted isoquinoline (**6a-b**) required for our reaction were prepared from isocoumarin^{8, 11} by following reported procedures¹⁶⁻¹⁸. Purified 1-hydrazino-3-substituted isoquinoline derivatives were allowed to react with 1,3-diketones (**7a-c**) in presence of anhydrous ethanol under nitrogen atmosphere to give the corresponding Isoquinolin-1-yl-2-(cycloalk-2-enylidene)hydrazines derivatives (**8a-f**), (Scheme 1; Table 1). The reaction producing hydrazine derivatives by a simple and an efficient route gave good yields. Products of the reaction have been isolated, purified and characterized by various spectral techniques such as IR, LC-MS, ¹H-NMR and ¹³C-NMR techniques.

Table 2 Antibacterial and Antifungal Activities of Compounds **8a-f**

Organisms	Titled compounds						Standards (30mcg/disc)
	8a	8b	8c	8d	8e	8f	
<i>Antibacterial activity</i>							
<i>E. coli</i>	-	-	-	-	-	-	23 (Ca ³⁰)
<i>S. typhi</i>	-	-	-	-	-	-	28 (C ³⁰)
<i>P. mirabilis</i>	-	-	-	-	-	-	20 (Ca ³⁰)
<i>B. cerus</i>	+	+	+	-	-	-	20 (Ca ³⁰)
<i>S. aureus</i>	-	-	-	-	-	-	22 (Na ³⁰)
<i>Antifungal activity</i>							
<i>C. albicans</i>	-	-	-	++	++	++	19 (Na ³⁰)
<i>A. flavus</i>	-	-	-	-	-	-	-
<i>A. niger</i>	-	-	-	-	-	-	-

Ca³⁰ - Ceftazidime, C³⁰ - Chloramphenicol, Na³⁰ - Nalidixic acid, Control = DMSO, + = 12 to 13 mm, ++ = 14 to 15 mm

LCMS spectra

The LCMS of titled compounds showed a molecular ion peak M+ in the positive mode. The molecular ion peak for **8a** was observed at m/z = 316.12. This is also supported by the LCMS analysis of other compounds, **8b-8f**.

IR and NMR spectra

In IR spectrum, reactant diketones **7a-c** gave peaks at around 1730-1710 cm^{-1} indicates the presence of carbonyl groups where as in products **8a-f** obtained from **7a-c** gave peaks at around 3300-3200 cm^{-1} indicates the reaction between **7a-c** and **6a-f** with the formation of an enolic group from an unreacted keto group of diketones.

In ^1H NMR spectra, peaks at δ 9.55- 9.75 ppm indicates the presence of an enolic hydroxyl group. Similarly ^{13}C NMR peaks at around δ 190- 205 ppm indicates enolic carbon thereby confirming isomerism of an unreacted keto group of diketone to enol form.

Antimicrobial studies

Among various Isoquinolin-1-yl-2-(cycloalk-2-enylidene)hydrazines, those with electron rich groups present in the compounds (**8a-c**) possess good antibacterial activity against *B. cerus* as well as those having electron deficient groups present in the compounds (**8d-f**) possess good antifungal activity against *C. albicans*. From the screening data given in Table 2, it is evident that compounds **8a**, **8b** and **8c** exhibited the highest degree of inhibition only against bacterial species *B. cerus*, compounds **8d**, **8e** and **8f** showed highest degree of inhibition only against fungal species *C. Albicans*. However, the activities of tested compounds are less than that of the standard agent used.

CONCLUSIONS

We have described a simple method for the synthesis of new heterocyclic Schiff bases by using 1, 3 cyclic diketones and hydrazinoisoquinoline derivatives. The method offers several advantages including high yields and simple work up procedure for the transformation of 1-hydrazino-3-substituted isoquinoline into isoquinolin-1-yl-2-(cycloalk-2-enylidene) hydrazine. The antimicrobial activities including antibacterial and antifungal properties of the synthesized compounds showed the titled compounds as biologically valuable materials.

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