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## Structure Reports

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## 2-[5-(4-Methoxyphenyl)-3-phenyl-4,5-dihydro-1H-pyrazol-1-yl]-6-methyl-1,3-benzothiazole

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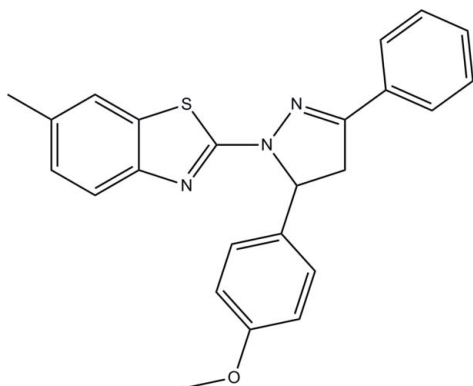
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Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.041;  $wR$  factor = 0.123; data-to-parameter ratio = 22.5.

In the title compound,  $\text{C}_{24}\text{H}_{21}\text{N}_3\text{OS}$ , the pyrazole ring makes dihedral angles of  $5.40$  (7) and  $6.72$  (8)° with the benzo[*d*]thiazole ring system and the benzene ring, respectively, and a dihedral angle of  $85.72$  (8)° with the methoxy-substituted benzene ring. In the crystal structure, the molecules are linked by  $\text{C}-\text{H}\cdots\pi$  interactions.

### Related literature

For background to the properties and applications of pyrazolines, see: Taylor *et al.* (1992); Rajendera Prasad *et al.* (2005). For reference bond-length data, see: Allen *et al.* (1987).



### Experimental

#### Crystal data

 $\text{C}_{24}\text{H}_{21}\text{N}_3\text{OS}$  $M_r = 399.50$ Orthorhombic, *Pbcn*

$a = 22.632$  (3) Å  
 $b = 11.1961$  (12) Å  
 $c = 16.1137$  (18) Å  
 $V = 4083.1$  (8) Å<sup>3</sup>

 $Z = 8$ Mo  $K\alpha$  radiation $\mu = 0.18$  mm<sup>-1</sup> $T = 296$  K $0.37 \times 0.24 \times 0.19$  mm

#### Data collection

Bruker SMART APEXII DUO  
CCD diffractometer  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2009)  
 $T_{\min} = 0.936$ ,  $T_{\max} = 0.967$

25081 measured reflections  
5949 independent reflections  
3835 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.041$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$   
 $wR(F^2) = 0.123$   
 $S = 1.00$   
5949 reflections

264 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.25$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.29$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$C_{g1}$ ,  $C_{g2}$  and  $C_{g3}$  are the centroids of the  $S1/C17/N1/C18/C23$ ,  $C1-C6$  and  $C10-C15$  rings, respectively.

| $D-H\cdots A$              | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|----------------------------|-------|-------------|-------------|---------------|
| $C15-H15A\cdots C_{g1}^i$  | 0.93  | 2.91        | 3.6318 (17) | 138           |
| $C22-H22A\cdots C_{g2}^i$  | 0.93  | 2.89        | 3.6438 (18) | 140           |
| $C2-H2A\cdots C_{g3}^{ii}$ | 0.93  | 2.74        | 3.4884 (18) | 138           |

Symmetry codes: (i)  $x + \frac{3}{2}, -y + \frac{1}{2}, -z$ ; (ii)  $x, -y - 1, z - \frac{1}{2}$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6368).

### References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.  
Bruker (2009). *SADABS*, *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Rajendera Prasad, Y., Lakshmana Rao, A., Prasoona, L., Murali, K. & Ravi Kumar, P. (2005). *Bioorg. Med. Chem. Lett.* **15**, 5030–5034.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.  
Taylor, E. C., Patel, H. & Kumar, H. (1992). *Tetrahedron*, **48**, 8089–8100.

\* Thomson Reuters ResearcherID: A-3561-2009.

**supplementary materials**

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## 2-[5-(4-Methoxyphenyl)-3-phenyl-4,5-dihydro-1*H*-pyrazol-1-yl]-6-methyl-1,3-benzothiazole

H.-K. Fun, S. Arshad, M. Himaja, D. Munirajasekhar and B. K. Sarojini

### Comment

Pyrazolines are an important class of heterocyclic compounds, some of which exhibit important pharmacological activities such as antitumor (Taylor *et al.*, 1992) and antidepressant (Rajendera Prasad *et al.*, 2005) agents. The title compound, (I), was synthesized by the condensation of 1-(6-methylbenzo[*d*]thiazol-2-yl)hydrazine with (*E*)-3-(4-methoxyphenyl)-1-phenylprop-2-en-1-one in presence of ethanol and its crystal structure is now described.

In the molecular structure (Fig 1), the pyrazole ring (N2/N3/C7–C9) is approximately planar with the benzo[*d*]thiazole ring system (S1/N1/C17–C23) and the benzene ring (C1–C6) with dihedral angles of 5.40 (7)° and 6.72 (8)°, respectively. On the other hand, the pyrazole ring (N2/N3/C7–C9) is approximately perpendicular to the methoxy substituted benzene ring (C10–C15) with dihedral angle of 85.72 (8)°. The bond lengths (Allen *et al.*, 1987) and angles are within normal ranges.

The crystal packing is shown in Fig. 2. The crystal structure is stabilized by the intermolecular C15–H15A...Cg1, C22—H22A...Cg2 and C2—H2A...Cg3 (Table 1) interactions (Cg1, Cg2 and Cg3 are the centroids of S1/C17/N1/C18/C23, C1—C6 and C10—C15 rings, respectively).

### Experimental

A mixture of (*E*)-3-(4-methoxyphenyl)-1-phenylprop-2-en-1-one (5 mmol) and 1-(6-methylbenzo[*d*]thiazol-2-yl)hydrazine (5 mmol) was refluxed for 16 h in ethanol (20 ml). After completion of the reaction, the reaction mixture was poured into cold water. The precipitate obtained was filtered and washed with cold water. The product was recrystallized from ethanol to yield colourless blocks of (I) and dried. m.p. 131–132 °C, HRMS Calcd for C<sub>24</sub>H<sub>21</sub>N<sub>3</sub>OS 399.5080 found 399.5079.

### Refinement

All H atoms were positioned geometrically [C—H = 0.93–0.98 Å] and refined using a riding model with  $U_{\text{iso}}(\text{H}) = 1.2$  or 1.5  $U_{\text{eq}}(\text{C})$ . A rotating group model was applied to the methyl groups.

### Figures

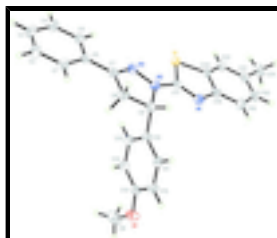


Fig. 1. The molecular structure of the title compound, showing 30% probability displacement ellipsoids.

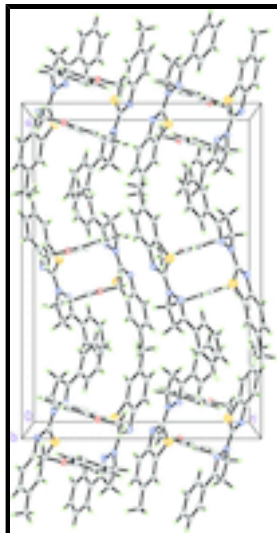


Fig. 2. The crystal packing of the title compound.

**2-[5-(4-Methoxyphenyl)-3-phenyl-4,5-dihydro-1H-pyrazol-1-yl]-6-methyl- 1,3-benzothiazole**

*Crystal data*

$C_{24}H_{21}N_3OS$

$M_r = 399.50$

Orthorhombic, *Pbcn*

Hall symbol: -P 2n 2ab

$a = 22.632 (3) \text{ \AA}$

$b = 11.1961 (12) \text{ \AA}$

$c = 16.1137 (18) \text{ \AA}$

$V = 4083.1 (8) \text{ \AA}^3$

$Z = 8$

$F(000) = 1680$

$D_x = 1.300 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4472 reflections

$\theta = 2.4\text{--}27.1^\circ$

$\mu = 0.18 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Block, colourless

$0.37 \times 0.24 \times 0.19 \text{ mm}$

*Data collection*

Bruker SMART APEXII DUO CCD diffractometer

Radiation source: fine-focus sealed tube graphite

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan (*SADABS*; Bruker, 2009)

$T_{\min} = 0.936$ ,  $T_{\max} = 0.967$

25081 measured reflections

5949 independent reflections

3835 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.041$

$\theta_{\max} = 30.0^\circ$ ,  $\theta_{\min} = 1.8^\circ$

$h = -31 \rightarrow 31$

$k = -15 \rightarrow 10$

$l = -22 \rightarrow 18$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

|                                 |  |
|---------------------------------|--|
| $R[F^2 > 2\sigma(F^2)] = 0.041$ | Hydrogen site location: inferred from neighbouring sites |
| $wR(F^2) = 0.123$               | H-atom parameters constrained                            |
| $S = 1.00$                      | $w = 1/[\sigma^2(F_o^2) + (0.0593P)^2 + 0.3403P]$        |
| 5949 reflections                | where $P = (F_o^2 + 2F_c^2)/3$                           |
| 264 parameters                  | $(\Delta/\sigma)_{\max} = 0.002$                         |
| 0 restraints                    | $\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$    |
|                                 | $\Delta\rho_{\min} = -0.29 \text{ e } \text{\AA}^{-3}$   |

### Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|     | $x$           | $y$           | $z$          | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|---------------|---------------|--------------|----------------------------------|
| S1  | 0.537594 (17) | -0.00428 (3)  | 0.11748 (3)  | 0.04063 (11)                     |
| N1  | 0.51267 (5)   | 0.20397 (10)  | 0.05137 (8)  | 0.0394 (3)                       |
| N2  | 0.43376 (5)   | 0.11316 (10)  | 0.11985 (8)  | 0.0408 (3)                       |
| N3  | 0.41278 (5)   | 0.01631 (10)  | 0.16416 (8)  | 0.0382 (3)                       |
| O1  | 0.43721 (6)   | 0.61031 (10)  | 0.32050 (8)  | 0.0589 (3)                       |
| C1  | 0.26306 (7)   | -0.02592 (14) | 0.24336 (11) | 0.0463 (4)                       |
| H1A | 0.2459        | 0.0437        | 0.2231       | 0.056*                           |
| C2  | 0.22952 (7)   | -0.10444 (16) | 0.29063 (11) | 0.0529 (4)                       |
| H2A | 0.1902        | -0.0867       | 0.3023       | 0.063*                           |
| C3  | 0.25414 (8)   | -0.20806 (16) | 0.32013 (11) | 0.0537 (4)                       |
| H3A | 0.2315        | -0.2608       | 0.3514       | 0.064*                           |
| C4  | 0.31269 (8)   | -0.23390 (15) | 0.30326 (11) | 0.0535 (4)                       |
| H4A | 0.3293        | -0.3041       | 0.3235       | 0.064*                           |
| C5  | 0.34660 (7)   | -0.15656 (13) | 0.25674 (10) | 0.0450 (4)                       |
| H5A | 0.3859        | -0.1748       | 0.2456       | 0.054*                           |
| C6  | 0.32201 (6)   | -0.05056 (12) | 0.22615 (9)  | 0.0378 (3)                       |
| C7  | 0.35724 (6)   | 0.03383 (12)  | 0.17774 (10) | 0.0377 (3)                       |
| C8  | 0.33397 (7)   | 0.14714 (13)  | 0.13926 (10) | 0.0427 (4)                       |
| H8A | 0.3117        | 0.1938        | 0.1792       | 0.051*                           |
| H8B | 0.3091        | 0.1301        | 0.0917       | 0.051*                           |
| C9  | 0.39093 (6)   | 0.21231 (12)  | 0.11292 (10) | 0.0384 (3)                       |
| H9A | 0.3879        | 0.2390        | 0.0552       | 0.046*                           |
| C10 | 0.40585 (6)   | 0.31648 (12)  | 0.16894 (9)  | 0.0340 (3)                       |
| C11 | 0.39769 (7)   | 0.43310 (13)  | 0.14186 (10) | 0.0423 (4)                       |

## supplementary materials

|      |              |              |               |            |
|------|--------------|--------------|---------------|------------|
| H11A | 0.3851       | 0.4470       | 0.0878        | 0.051*     |
| C12  | 0.40797 (8)  | 0.52875 (14) | 0.19391 (10)  | 0.0471 (4) |
| H12A | 0.4020       | 0.6062       | 0.1749        | 0.056*     |
| C13  | 0.42714 (6)  | 0.50954 (13) | 0.27438 (10)  | 0.0396 (3) |
| C14  | 0.43514 (7)  | 0.39440 (13) | 0.30260 (10)  | 0.0411 (3) |
| H14A | 0.4477       | 0.3807       | 0.3567        | 0.049*     |
| C15  | 0.42441 (7)  | 0.29915 (13) | 0.24975 (10)  | 0.0399 (3) |
| H15A | 0.4298       | 0.2217       | 0.2691        | 0.048*     |
| C16  | 0.45090 (10) | 0.5966 (2)   | 0.40532 (13)  | 0.0768 (6) |
| H16A | 0.4569       | 0.6737       | 0.4299        | 0.115*     |
| H16B | 0.4863       | 0.5499       | 0.4109        | 0.115*     |
| H16C | 0.4189       | 0.5567       | 0.4329        | 0.115*     |
| C17  | 0.49059 (6)  | 0.11551 (12) | 0.09394 (9)   | 0.0348 (3) |
| C18  | 0.57140 (6)  | 0.17917 (13) | 0.03344 (9)   | 0.0388 (3) |
| C19  | 0.60917 (8)  | 0.25405 (16) | -0.01089 (10) | 0.0508 (4) |
| H19A | 0.5956       | 0.3265       | -0.0317       | 0.061*     |
| C20  | 0.66700 (8)  | 0.21907 (17) | -0.02351 (11) | 0.0550 (4) |
| H20A | 0.6919       | 0.2688       | -0.0538       | 0.066*     |
| C21  | 0.68940 (7)  | 0.11232 (16) | 0.00738 (10)  | 0.0498 (4) |
| C22  | 0.65199 (7)  | 0.03749 (15) | 0.05197 (10)  | 0.0455 (4) |
| H22A | 0.6660       | -0.0343      | 0.0734        | 0.055*     |
| C23  | 0.59353 (6)  | 0.07106 (13) | 0.06414 (9)   | 0.0375 (3) |
| C24  | 0.75341 (8)  | 0.0795 (2)   | -0.00596 (13) | 0.0688 (5) |
| H24A | 0.7616       | 0.0046       | 0.0207        | 0.103*     |
| H24B | 0.7782       | 0.1405       | 0.0173        | 0.103*     |
| H24C | 0.7611       | 0.0727       | -0.0644       | 0.103*     |

### Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$    | $U^{22}$     | $U^{33}$    | $U^{12}$      | $U^{13}$     | $U^{23}$     |
|-----|-------------|--------------|-------------|---------------|--------------|--------------|
| S1  | 0.0411 (2)  | 0.03568 (19) | 0.0451 (2)  | -0.00003 (15) | 0.00495 (16) | 0.00259 (15) |
| N1  | 0.0411 (7)  | 0.0400 (7)   | 0.0372 (7)  | -0.0024 (5)   | 0.0005 (5)   | 0.0015 (5)   |
| N2  | 0.0386 (7)  | 0.0320 (6)   | 0.0519 (8)  | -0.0002 (5)   | 0.0072 (6)   | 0.0028 (5)   |
| N3  | 0.0376 (6)  | 0.0334 (6)   | 0.0435 (8)  | -0.0027 (5)   | 0.0044 (5)   | -0.0023 (5)  |
| O1  | 0.0692 (8)  | 0.0499 (7)   | 0.0575 (8)  | 0.0019 (6)    | -0.0072 (6)  | -0.0192 (6)  |
| C1  | 0.0370 (7)  | 0.0434 (8)   | 0.0585 (11) | -0.0011 (6)   | -0.0023 (7)  | -0.0003 (7)  |
| C2  | 0.0336 (8)  | 0.0619 (11)  | 0.0631 (12) | -0.0049 (7)   | 0.0055 (7)   | -0.0002 (9)  |
| C3  | 0.0483 (9)  | 0.0560 (10)  | 0.0568 (11) | -0.0135 (8)   | 0.0059 (8)   | 0.0069 (8)   |
| C4  | 0.0527 (10) | 0.0435 (9)   | 0.0642 (12) | -0.0015 (7)   | 0.0048 (8)   | 0.0091 (8)   |
| C5  | 0.0399 (8)  | 0.0400 (8)   | 0.0551 (10) | 0.0008 (6)    | 0.0041 (7)   | -0.0010 (7)  |
| C6  | 0.0353 (7)  | 0.0369 (7)   | 0.0411 (9)  | -0.0048 (6)   | -0.0001 (6)  | -0.0053 (6)  |
| C7  | 0.0359 (7)  | 0.0348 (7)   | 0.0423 (9)  | -0.0030 (6)   | -0.0017 (6)  | -0.0057 (6)  |
| C8  | 0.0356 (7)  | 0.0377 (8)   | 0.0548 (10) | -0.0020 (6)   | -0.0040 (7)  | -0.0013 (7)  |
| C9  | 0.0388 (7)  | 0.0360 (7)   | 0.0405 (8)  | 0.0002 (6)    | -0.0031 (6)  | 0.0009 (6)   |
| C10 | 0.0311 (7)  | 0.0339 (7)   | 0.0369 (8)  | 0.0001 (5)    | -0.0003 (6)  | 0.0027 (6)   |
| C11 | 0.0536 (9)  | 0.0375 (8)   | 0.0358 (8)  | 0.0010 (6)    | -0.0030 (7)  | 0.0059 (6)   |
| C12 | 0.0623 (10) | 0.0324 (7)   | 0.0465 (10) | 0.0029 (7)    | 0.0002 (8)   | 0.0036 (6)   |
| C13 | 0.0337 (7)  | 0.0419 (8)   | 0.0434 (9)  | 0.0010 (6)    | 0.0013 (6)   | -0.0056 (7)  |

|     |             |             |             |              |              |              |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C14 | 0.0370 (7)  | 0.0487 (8)  | 0.0374 (8)  | 0.0031 (6)   | -0.0052 (6)  | 0.0008 (6)   |
| C15 | 0.0399 (8)  | 0.0372 (7)  | 0.0425 (9)  | 0.0025 (6)   | -0.0052 (6)  | 0.0062 (6)   |
| C16 | 0.0759 (14) | 0.0946 (16) | 0.0600 (13) | 0.0059 (12)  | -0.0087 (11) | -0.0341 (12) |
| C17 | 0.0376 (7)  | 0.0346 (7)  | 0.0322 (8)  | -0.0022 (6)  | -0.0009 (6)  | -0.0048 (6)  |
| C18 | 0.0409 (8)  | 0.0436 (8)  | 0.0319 (8)  | -0.0050 (6)  | 0.0007 (6)   | -0.0028 (6)  |
| C19 | 0.0548 (10) | 0.0510 (10) | 0.0466 (10) | -0.0072 (7)  | 0.0057 (8)   | 0.0075 (8)   |
| C20 | 0.0513 (10) | 0.0659 (11) | 0.0477 (10) | -0.0166 (8)  | 0.0107 (8)   | 0.0010 (8)   |
| C21 | 0.0401 (8)  | 0.0671 (11) | 0.0422 (9)  | -0.0069 (8)  | 0.0043 (7)   | -0.0114 (8)  |
| C22 | 0.0420 (8)  | 0.0516 (9)  | 0.0429 (9)  | 0.0011 (7)   | 0.0006 (7)   | -0.0048 (7)  |
| C23 | 0.0399 (7)  | 0.0413 (8)  | 0.0314 (8)  | -0.0042 (6)  | 0.0021 (6)   | -0.0019 (6)  |
| C24 | 0.0437 (9)  | 0.0941 (15) | 0.0686 (13) | -0.0049 (10) | 0.0110 (9)   | -0.0166 (11) |

*Geometric parameters (Å, °)*

|            |             |              |             |
|------------|-------------|--------------|-------------|
| S1—C23     | 1.7473 (15) | C9—H9A       | 0.9800      |
| S1—C17     | 1.7534 (15) | C10—C15      | 1.382 (2)   |
| N1—C17     | 1.3042 (18) | C10—C11      | 1.3891 (19) |
| N1—C18     | 1.3884 (19) | C11—C12      | 1.380 (2)   |
| N2—C17     | 1.3525 (18) | C11—H11A     | 0.9300      |
| N2—N3      | 1.3824 (16) | C12—C13      | 1.384 (2)   |
| N2—C9      | 1.4780 (18) | C12—H12A     | 0.9300      |
| N3—C7      | 1.2909 (18) | C13—C14      | 1.379 (2)   |
| O1—C13     | 1.3701 (17) | C14—C15      | 1.386 (2)   |
| O1—C16     | 1.410 (2)   | C14—H14A     | 0.9300      |
| C1—C2      | 1.389 (2)   | C15—H15A     | 0.9300      |
| C1—C6      | 1.390 (2)   | C16—H16A     | 0.9600      |
| C1—H1A     | 0.9300      | C16—H16B     | 0.9600      |
| C2—C3      | 1.372 (2)   | C16—H16C     | 0.9600      |
| C2—H2A     | 0.9300      | C18—C19      | 1.394 (2)   |
| C3—C4      | 1.383 (2)   | C18—C23      | 1.400 (2)   |
| C3—H3A     | 0.9300      | C19—C20      | 1.381 (2)   |
| C4—C5      | 1.379 (2)   | C19—H19A     | 0.9300      |
| C4—H4A     | 0.9300      | C20—C21      | 1.390 (3)   |
| C5—C6      | 1.400 (2)   | C20—H20A     | 0.9300      |
| C5—H5A     | 0.9300      | C21—C22      | 1.391 (2)   |
| C6—C7      | 1.462 (2)   | C21—C24      | 1.510 (2)   |
| C7—C8      | 1.507 (2)   | C22—C23      | 1.389 (2)   |
| C8—C9      | 1.541 (2)   | C22—H22A     | 0.9300      |
| C8—H8A     | 0.9700      | C24—H24A     | 0.9600      |
| C8—H8B     | 0.9700      | C24—H24B     | 0.9600      |
| C9—C10     | 1.513 (2)   | C24—H24C     | 0.9600      |
| C23—S1—C17 | 87.93 (7)   | C11—C12—C13  | 120.11 (14) |
| C17—N1—C18 | 108.92 (12) | C11—C12—H12A | 119.9       |
| C17—N2—N3  | 120.09 (11) | C13—C12—H12A | 119.9       |
| C17—N2—C9  | 125.85 (12) | O1—C13—C14   | 124.69 (14) |
| N3—N2—C9   | 113.76 (11) | O1—C13—C12   | 115.62 (13) |
| C7—N3—N2   | 107.63 (12) | C14—C13—C12  | 119.69 (13) |
| C13—O1—C16 | 118.20 (14) | C13—C14—C15  | 119.58 (14) |
| C2—C1—C6   | 120.55 (15) | C13—C14—H14A | 120.2       |

## supplementary materials

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|              |              |                 |              |
|--------------|--------------|-----------------|--------------|
| C2—C1—H1A    | 119.7        | C15—C14—H14A    | 120.2        |
| C6—C1—H1A    | 119.7        | C10—C15—C14     | 121.61 (13)  |
| C3—C2—C1     | 120.21 (15)  | C10—C15—H15A    | 119.2        |
| C3—C2—H2A    | 119.9        | C14—C15—H15A    | 119.2        |
| C1—C2—H2A    | 119.9        | O1—C16—H16A     | 109.5        |
| C2—C3—C4     | 119.85 (15)  | O1—C16—H16B     | 109.5        |
| C2—C3—H3A    | 120.1        | H16A—C16—H16B   | 109.5        |
| C4—C3—H3A    | 120.1        | O1—C16—H16C     | 109.5        |
| C5—C4—C3     | 120.58 (16)  | H16A—C16—H16C   | 109.5        |
| C5—C4—H4A    | 119.7        | H16B—C16—H16C   | 109.5        |
| C3—C4—H4A    | 119.7        | N1—C17—N2       | 122.78 (13)  |
| C4—C5—C6     | 120.16 (15)  | N1—C17—S1       | 117.53 (11)  |
| C4—C5—H5A    | 119.9        | N2—C17—S1       | 119.69 (10)  |
| C6—C5—H5A    | 119.9        | N1—C18—C19      | 124.98 (14)  |
| C1—C6—C5     | 118.64 (14)  | N1—C18—C23      | 116.23 (13)  |
| C1—C6—C7     | 120.11 (14)  | C19—C18—C23     | 118.78 (14)  |
| C5—C6—C7     | 121.25 (13)  | C20—C19—C18     | 119.07 (16)  |
| N3—C7—C6     | 121.56 (13)  | C20—C19—H19A    | 120.5        |
| N3—C7—C8     | 113.48 (13)  | C18—C19—H19A    | 120.5        |
| C6—C7—C8     | 124.96 (13)  | C19—C20—C21     | 122.44 (15)  |
| C7—C8—C9     | 102.69 (12)  | C19—C20—H20A    | 118.8        |
| C7—C8—H8A    | 111.2        | C21—C20—H20A    | 118.8        |
| C9—C8—H8A    | 111.2        | C20—C21—C22     | 118.73 (15)  |
| C7—C8—H8B    | 111.2        | C20—C21—C24     | 120.51 (16)  |
| C9—C8—H8B    | 111.2        | C22—C21—C24     | 120.74 (17)  |
| H8A—C8—H8B   | 109.1        | C23—C22—C21     | 119.32 (16)  |
| N2—C9—C10    | 112.80 (12)  | C23—C22—H22A    | 120.3        |
| N2—C9—C8     | 99.91 (11)   | C21—C22—H22A    | 120.3        |
| C10—C9—C8    | 112.79 (12)  | C22—C23—C18     | 121.65 (14)  |
| N2—C9—H9A    | 110.3        | C22—C23—S1      | 128.96 (12)  |
| C10—C9—H9A   | 110.3        | C18—C23—S1      | 109.38 (11)  |
| C8—C9—H9A    | 110.3        | C21—C24—H24A    | 109.5        |
| C15—C10—C11  | 117.93 (13)  | C21—C24—H24B    | 109.5        |
| C15—C10—C9   | 121.45 (12)  | H24A—C24—H24B   | 109.5        |
| C11—C10—C9   | 120.50 (13)  | C21—C24—H24C    | 109.5        |
| C12—C11—C10  | 121.07 (15)  | H24A—C24—H24C   | 109.5        |
| C12—C11—H11A | 119.5        | H24B—C24—H24C   | 109.5        |
| C10—C11—H11A | 119.5        |                 |              |
| C17—N2—N3—C7 | -177.07 (13) | C11—C12—C13—O1  | 178.73 (15)  |
| C9—N2—N3—C7  | 8.82 (17)    | C11—C12—C13—C14 | -0.9 (2)     |
| C6—C1—C2—C3  | -0.7 (3)     | O1—C13—C14—C15  | -178.95 (14) |
| C1—C2—C3—C4  | 0.5 (3)      | C12—C13—C14—C15 | 0.6 (2)      |
| C2—C3—C4—C5  | -0.2 (3)     | C11—C10—C15—C14 | -0.5 (2)     |
| C3—C4—C5—C6  | 0.1 (3)      | C9—C10—C15—C14  | -176.56 (13) |
| C2—C1—C6—C5  | 0.6 (2)      | C13—C14—C15—C10 | 0.1 (2)      |
| C2—C1—C6—C7  | -178.78 (15) | C18—N1—C17—N2   | -179.24 (13) |
| C4—C5—C6—C1  | -0.3 (2)     | C18—N1—C17—S1   | 0.64 (16)    |
| C4—C5—C6—C7  | 179.05 (15)  | N3—N2—C17—N1    | 179.42 (13)  |
| N2—N3—C7—C6  | -177.70 (13) | C9—N2—C17—N1    | -7.2 (2)     |



|                 |              |                 |              |
|-----------------|--------------|-----------------|--------------|
| N2—N3—C7—C8     | 2.31 (17)    | N3—N2—C17—S1    | -0.46 (18)   |
| C1—C6—C7—N3     | 176.59 (15)  | C9—N2—C17—S1    | 172.88 (11)  |
| C5—C6—C7—N3     | -2.8 (2)     | C23—S1—C17—N1   | -0.27 (12)   |
| C1—C6—C7—C8     | -3.4 (2)     | C23—S1—C17—N2   | 179.61 (12)  |
| C5—C6—C7—C8     | 177.19 (14)  | C17—N1—C18—C19  | -179.67 (15) |
| N3—C7—C8—C9     | -11.49 (17)  | C17—N1—C18—C23  | -0.79 (18)   |
| C6—C7—C8—C9     | 168.52 (13)  | N1—C18—C19—C20  | 179.20 (15)  |
| C17—N2—C9—C10   | -68.81 (19)  | C23—C18—C19—C20 | 0.3 (2)      |
| N3—N2—C9—C10    | 104.90 (14)  | C18—C19—C20—C21 | -0.9 (3)     |
| C17—N2—C9—C8    | 171.19 (14)  | C19—C20—C21—C22 | 0.6 (3)      |
| N3—N2—C9—C8     | -15.11 (16)  | C19—C20—C21—C24 | -178.27 (17) |
| C7—C8—C9—N2     | 14.51 (15)   | C20—C21—C22—C23 | 0.2 (2)      |
| C7—C8—C9—C10    | -105.50 (14) | C24—C21—C22—C23 | 179.10 (15)  |
| N2—C9—C10—C15   | -42.67 (19)  | C21—C22—C23—C18 | -0.8 (2)     |
| C8—C9—C10—C15   | 69.62 (17)   | C21—C22—C23—S1  | -179.64 (12) |
| N2—C9—C10—C11   | 141.35 (14)  | N1—C18—C23—C22  | -178.48 (13) |
| C8—C9—C10—C11   | -106.37 (16) | C19—C18—C23—C22 | 0.5 (2)      |
| C15—C10—C11—C12 | 0.2 (2)      | N1—C18—C23—S1   | 0.60 (16)    |
| C9—C10—C11—C12  | 176.33 (14)  | C19—C18—C23—S1  | 179.55 (12)  |
| C10—C11—C12—C13 | 0.5 (2)      | C17—S1—C23—C22  | 178.81 (15)  |
| C16—O1—C13—C14  | -6.9 (2)     | C17—S1—C23—C18  | -0.18 (11)   |
| C16—O1—C13—C12  | 173.59 (16)  |                 |              |

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

Cg1, Cg2 and Cg3 are the centroids of the S1/C17/N1/C18/C23, C1—C6 and C10—C15 rings, respectively.

| $D-H\cdots A$                      | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------------|-------|-------------|-------------|---------------|
| C15—H15A $\cdots$ Cg1 <sup>i</sup> | 0.93  | 2.91        | 3.6318 (17) | 138           |
| C22—H22A $\cdots$ Cg2 <sup>i</sup> | 0.93  | 2.89        | 3.6438 (18) | 140           |
| C2—H2A $\cdots$ Cg3 <sup>ii</sup>  | 0.93  | 2.74        | 3.4884 (18) | 138           |

Symmetry codes: (i)  $x+3/2, -y+1/2, -z$ ; (ii)  $x, -y-1, z-1/2$ .

Fig. 1

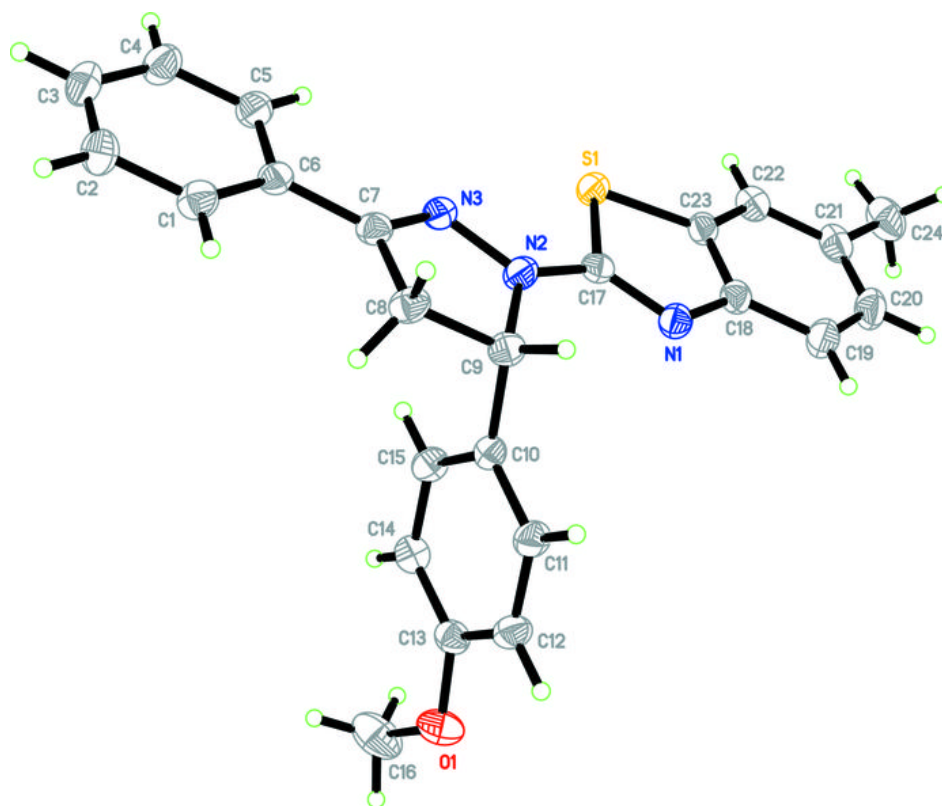
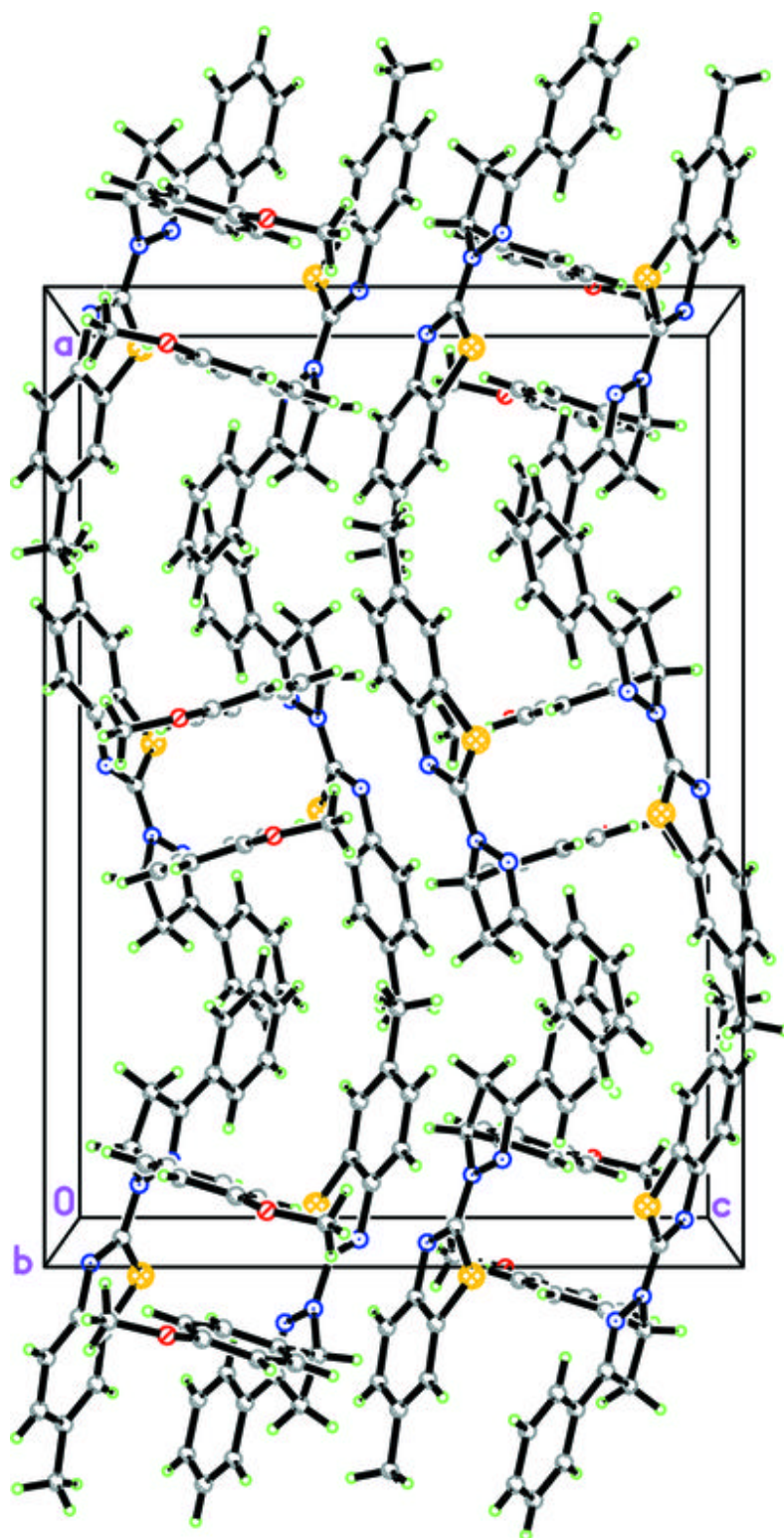


Fig. 2



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