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1-[(2-Chloro-3-quinolyl)methyl]indoline-2,3-dione

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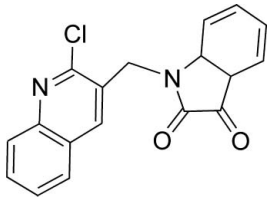
Key indicators: single-crystal X-ray study; T = 293 K; mean σ(C–C) = 0.003 Å; R factor = 0.037; wR factor = 0.085; data-to-parameter ratio = 13.1.

In the title compound, $C_{18}H_{11}ClN_2O_2$, the isatin and 2-chloro-3-methylquinoline units are both almost planar, with r.m.s. deviations of 0.0075 and 0.0086 Å, respectively, and the dihedral angle between the mean planes of the two units is 83.13 (7)°. In the crystal, a weak intermolecular C—H$\cdots$O interaction links the molecules into chains along the c axis.

Related literature

For background to the use of *N*-substituted indole-2,3-diones as intermediates and synthetic precursors for the preparation of heterocyclic compounds, see: Silaicheva *et al.* (2009). For the biological activity of *N*-substituted indole-2,3-diones, see: Vine *et al.* (2007). For reference bond lengths, see: Allen *et al.* (1987).



Experimental

Crystal data

$C_{18}H_{11}ClN_2O_2$ M_r = 322.74
Monoclinic, $P2_1/c$
a = 21.4984 (8) Å
b = 5.3061 (2) Å
c = 13.0356 (4) Å
β = 99.718 (3)°
V = 1465.67 (9) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.27 mm^{-1}
T = 293 K
0.33 × 0.30 × 0.27 mm

Data collection

Oxford Diffraction Excalibur diffractometer
17920 measured reflections
2724 independent reflections
1731 reflections with $I > 2\sigma(I)$
R_{int} = 0.050

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.037
$wR(F^2)$ = 0.085
S = 0.90
2724 reflections
208 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.21 e Å^{-3}
$\Delta\rho_{min}$ = −0.23 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H$\cdots A$	D—H	H$\cdots A$	$D\cdots A$	D—H$\cdots A$
C17—H17$\cdots$O2^{i}	0.93	2.48	3.367 (3)	160

Symmetry code: (i) $x, -y+\frac{3}{2}, z+\frac{1}{2}$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PV2276).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Oxford Diffraction (2009). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, England.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Silaicheva, P. S., Alievb, Z. G. & Maslivetsa, A. N. (2009). *Russ. J. Org. Chem.* **45**, 126–130.
Vine, K. L., Locke, J. M., Ranson, M., Pyne, S. G. & Bremner, J. B. (2007). *J. Med. Chem.* **50**, 5109–5117.

supplementary materials

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1-[(2-Chloro-3-quinolyl)methyl]indoline-2,3-dione

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Comment

N-substituted indole-2,3-diones have been frequently used as intermediates and synthetic precursors for the preparation of a wide variety of heterocyclic compounds (Silaicheva *et al.*, 2009). In addition, they possess different biological activities such as cytotoxicity, antiviral activity and selective caspase inhibitions, etc. (Vine *et al.*, 2007). We have synthesized a novel isatin derivative and determined its crystal structure which is presented in this article.

In the title molecule, the atoms (C11—C18/N2/O1/O2) of the isatin moiety and 2-chloro-3-methylquinoline group (C1—C8/N1/Cl1) are individually planar with maximum r.m.s. deviations of 0.0075 and 0.0086 Å, respectively, from their mean-planes. The dihedral angle between the two ring systems is 83.13 (7)°. The bond distances and angles in the title compound are as expected (Allen *et al.*, 1987). There is a weak intermolecular interaction C17—H17··· O2 linking the molecules into chains along the *c*-axis.

Experimental

2-Chloro-3-chloromethylquinoline (210 mg, 1 mmol), KO^{t}Bu (112 mg, 1 mmol) and isatin (147 mg, 1 mmol) in tetrahydrofuran (10 ml) were taken in a round bottemed flask and the mixture was refluxed at 70 W for 3 min. Ethylacetate (30 ml) was poured into the reaction mixture and filtered off. The filtrate was subjected to column chromatography packed with silica and ethyl acetate/petroleum ether was used as the eluant (4:1). Crystals of suitable quality were grown by slow evaporation from a solution of the title compound in dichloromethane.

Refinement

Hydrogen atoms were placed in calculated positions at C—H = 0.93 and 0.97 Å, for aryl and methylene type H atoms, respectively, and were included in the refinement in riding model approximation, with U_{iso}(H) set to 1.2U_{eq}(C).

Figures

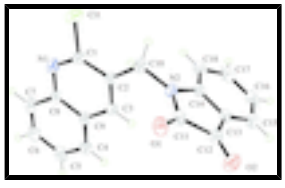

Fig. 1. Molecular structure of the title compound showing atom numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

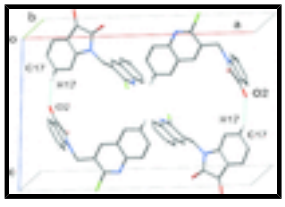

Fig. 2. Perspective view of the molecular packing of the title compound in the unit cell down the *b*-axis.

1-[(2-Chloro-3-quinolyl)methyl]indoline-2,3-dione

Crystal data

$C_{18}H_{11}ClN_2O_2$	$F(000) = 664$
$M_r = 322.74$	$D_x = 1.463$ Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: -P 2ybc	Cell parameters from 2724 reflections
$a = 21.4984$ (8) Å	$\theta = 2.9$–$25.5°$
$b = 5.3061$ (2) Å	$\mu = 0.27$ mm^{-1}
$c = 13.0356$ (4) Å	$T = 293$ K
$\beta = 99.718$ (3)°	Block, orange
$V = 1465.67$ (9) Å^3	0.33 × 0.30 × 0.27 mm
$Z = 4$	

Data collection

Oxford Diffraction Excalibur diffractometer	1731 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{int} = 0.050$
graphite	$\theta_{max} = 25.5°$, $\theta_{min} = 2.9°$
Detector resolution: 0 pixels mm^{-1}	$h = -26\rightarrow26$
ω scans	$k = -6\rightarrow6$
17920 measured reflections	$l = -15\rightarrow15$
2724 independent reflections	

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.037$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.085$	H-atom parameters constrained
$S = 0.90$	$w = 1/[\sigma^2(F_o^2) + (0.0447P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
2724 reflections	$(\Delta/\sigma)_{max} < 0.001$
208 parameters	$\Delta\rho_{max} = 0.21$ e Å^{-3}
0 restraints	$\Delta\rho_{min} = -0.23$ e Å^{-3}

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor wR and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) etc. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Cl1	0.32510 (3)	1.40527 (10)	0.51979 (4)	0.0614 (2)
N1	0.38948 (7)	1.0493 (3)	0.45048 (11)	0.0458 (4)
C2	0.28707 (8)	1.1294 (3)	0.34507 (13)	0.0360 (4)
C3	0.29797 (8)	0.9460 (3)	0.27719 (13)	0.0379 (4)
H3	0.2675	0.9121	0.2192	0.046*
N2	0.18214 (7)	1.2157 (3)	0.24049 (11)	0.0398 (4)
C13	0.10081 (8)	1.0002 (3)	0.13973 (13)	0.0376 (4)
C1	0.33623 (9)	1.1692 (3)	0.43088 (13)	0.0406 (5)
C14	0.13853 (8)	1.0147 (3)	0.23746 (14)	0.0377 (4)
C8	0.39988 (8)	0.8632 (3)	0.38138 (14)	0.0407 (5)
O1	0.20473 (6)	1.5156 (3)	0.12348 (11)	0.0579 (4)
C4	0.36754 (9)	0.6131 (3)	0.22547 (15)	0.0481 (5)
H4	0.3382	0.5751	0.1666	0.058*
O2	0.10171 (7)	1.2586 (3)	−0.01339 (11)	0.0635 (4)
C10	0.22789 (9)	1.2877 (3)	0.33140 (14)	0.0463 (5)
H10A	0.2085	1.2717	0.3931	0.056*
H10B	0.2392	1.4632	0.3249	0.056*
C9	0.35465 (8)	0.8066 (3)	0.29322 (13)	0.0375 (4)
C15	0.05482 (9)	0.8170 (4)	0.11805 (15)	0.0477 (5)
H15	0.0300	0.8064	0.0525	0.057*
C12	0.12083 (8)	1.1985 (3)	0.07592 (15)	0.0423 (5)
C7	0.45672 (9)	0.7252 (4)	0.39988 (16)	0.0547 (6)
H7	0.4871	0.7616	0.4576	0.066*
C5	0.42260 (10)	0.4820 (4)	0.24572 (18)	0.0587 (6)
H5	0.4306	0.3541	0.2010	0.070*
C11	0.17522 (9)	1.3370 (4)	0.14680 (15)	0.0418 (5)
C6	0.46691 (10)	0.5389 (4)	0.33319 (19)	0.0588 (6)
H6	0.5043	0.4475	0.3463	0.071*
C18	0.13105 (9)	0.8492 (4)	0.31574 (15)	0.0476 (5)
H18	0.1561	0.8586	0.3812	0.057*
C16	0.04657 (9)	0.6503 (4)	0.19561 (18)	0.0542 (6)
H16	0.0156	0.5264	0.1829	0.065*
C17	0.08426 (10)	0.6670 (4)	0.29230 (18)	0.0552 (6)
H17	0.0781	0.5524	0.3436	0.066*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0765 (4)	0.0609 (4)	0.0442 (3)	−0.0082 (3)	0.0028 (3)	−0.0161 (2)

N1	0.0427 (10)	0.0504 (10)	0.0406 (9)	−0.0072 (8)	−0.0039 (8)	0.0046 (8)
C2	0.0363 (10)	0.0362 (10)	0.0346 (10)	−0.0046 (9)	0.0038 (8)	−0.0006 (8)
C3	0.0337 (10)	0.0422 (11)	0.0359 (10)	−0.0036 (9)	0.0000 (8)	0.0003 (9)
N2	0.0341 (9)	0.0400 (9)	0.0425 (10)	0.0007 (7)	−0.0016 (7)	−0.0043 (7)
C13	0.0320 (10)	0.0364 (10)	0.0436 (12)	0.0019 (9)	0.0040 (9)	0.0000 (9)
C1	0.0460 (12)	0.0415 (11)	0.0335 (11)	−0.0096 (10)	0.0042 (9)	−0.0009 (9)
C14	0.0340 (11)	0.0335 (10)	0.0461 (12)	0.0066 (9)	0.0078 (9)	−0.0017 (9)
C8	0.0319 (11)	0.0430 (11)	0.0457 (12)	−0.0052 (9)	0.0027 (9)	0.0079 (9)
O1	0.0489 (9)	0.0526 (9)	0.0719 (10)	−0.0123 (7)	0.0092 (7)	0.0042 (7)
C4	0.0427 (12)	0.0496 (12)	0.0520 (13)	0.0032 (10)	0.0078 (10)	−0.0040 (10)
O2	0.0690 (10)	0.0721 (10)	0.0457 (9)	−0.0133 (8)	−0.0013 (8)	0.0077 (7)
C10	0.0439 (12)	0.0458 (11)	0.0465 (12)	0.0017 (10)	0.0004 (10)	−0.0106 (9)
C9	0.0326 (11)	0.0405 (10)	0.0388 (11)	−0.0039 (9)	0.0045 (9)	0.0038 (9)
C15	0.0394 (12)	0.0460 (11)	0.0554 (13)	−0.0005 (10)	0.0008 (10)	−0.0040 (10)
C12	0.0402 (12)	0.0448 (11)	0.0410 (12)	0.0026 (9)	0.0045 (10)	−0.0006 (10)
C7	0.0337 (12)	0.0623 (14)	0.0632 (14)	−0.0041 (11)	−0.0062 (10)	0.0136 (12)
C5	0.0492 (14)	0.0546 (13)	0.0741 (16)	0.0050 (11)	0.0160 (12)	−0.0028 (11)
C11	0.0330 (11)	0.0405 (11)	0.0516 (13)	0.0026 (10)	0.0069 (9)	0.0001 (10)
C6	0.0395 (13)	0.0548 (14)	0.0824 (17)	0.0074 (11)	0.0107 (12)	0.0099 (12)
C18	0.0476 (13)	0.0477 (12)	0.0477 (12)	0.0142 (11)	0.0087 (10)	0.0045 (10)
C16	0.0413 (13)	0.0421 (12)	0.0810 (17)	−0.0023 (10)	0.0155 (12)	−0.0007 (12)
C17	0.0576 (14)	0.0415 (12)	0.0726 (16)	0.0104 (11)	0.0282 (13)	0.0122 (11)

Geometric parameters (Å, °)

Cl1—C1	1.7503 (18)	C4—C9	1.412 (2)
N1—C1	1.297 (2)	C4—H4	0.9300
N1—C8	1.380 (2)	O2—C12	1.210 (2)
C2—C3	1.362 (2)	C10—H10A	0.9700
C2—C1	1.419 (3)	C10—H10B	0.9700
C2—C10	1.510 (2)	C15—C16	1.377 (3)
C3—C9	1.411 (2)	C15—H15	0.9300
C3—H3	0.9300	C12—C11	1.548 (3)
N2—C11	1.366 (2)	C7—C6	1.358 (3)
N2—C14	1.416 (2)	C7—H7	0.9300
N2—C10	1.457 (2)	C5—C6	1.390 (3)
C13—C15	1.381 (2)	C5—H5	0.9300
C13—C14	1.392 (2)	C6—H6	0.9300
C13—C12	1.451 (3)	C18—C17	1.391 (3)
C14—C18	1.376 (2)	C18—H18	0.9300
C8—C9	1.407 (2)	C16—C17	1.381 (3)
C8—C7	1.410 (3)	C16—H16	0.9300
O1—C11	1.208 (2)	C17—H17	0.9300
C4—C5	1.360 (3)		
C1—N1—C8	117.22 (16)	C8—C9—C3	118.02 (17)
C3—C2—C1	115.56 (17)	C8—C9—C4	118.98 (17)
C3—C2—C10	123.70 (16)	C3—C9—C4	123.00 (17)
C1—C2—C10	120.73 (16)	C16—C15—C13	118.58 (18)
C2—C3—C9	121.17 (17)	C16—C15—H15	120.7

C2—C3—H3	119.4	C13—C15—H15	120.7
C9—C3—H3	119.4	O2—C12—C13	131.09 (18)
C11—N2—C14	110.98 (15)	O2—C12—C11	123.28 (17)
C11—N2—C10	124.07 (15)	C13—C12—C11	105.62 (16)
C14—N2—C10	124.94 (15)	C6—C7—C8	119.8 (2)
C15—C13—C14	120.82 (16)	C6—C7—H7	120.1
C15—C13—C12	131.64 (17)	C8—C7—H7	120.1
C14—C13—C12	107.54 (16)	C4—C5—C6	120.1 (2)
N1—C1—C2	126.69 (17)	C4—C5—H5	119.9
N1—C1—Cl1	115.82 (14)	C6—C5—H5	119.9
C2—C1—Cl1	117.49 (15)	O1—C11—N2	127.74 (18)
C18—C14—C13	121.36 (17)	O1—C11—C12	126.75 (18)
C18—C14—N2	128.29 (17)	N2—C11—C12	105.50 (16)
C13—C14—N2	110.35 (15)	C7—C6—C5	121.4 (2)
N1—C8—C9	121.33 (17)	C7—C6—H6	119.3
N1—C8—C7	119.42 (18)	C5—C6—H6	119.3
C9—C8—C7	119.24 (18)	C14—C18—C17	116.82 (19)
C5—C4—C9	120.46 (19)	C14—C18—H18	121.6
C5—C4—H4	119.8	C17—C18—H18	121.6
C9—C4—H4	119.8	C15—C16—C17	120.00 (19)
N2—C10—C2	112.95 (14)	C15—C16—H16	120.0
N2—C10—H10A	109.0	C17—C16—H16	120.0
C2—C10—H10A	109.0	C16—C17—C18	122.42 (19)
N2—C10—H10B	109.0	C16—C17—H17	118.8
C2—C10—H10B	109.0	C18—C17—H17	118.8
H10A—C10—H10B	107.8		
C1—C2—C3—C9	0.3 (2)	C2—C3—C9—C4	179.27 (16)
C10—C2—C3—C9	179.03 (16)	C5—C4—C9—C8	0.6 (3)
C8—N1—C1—C2	−0.1 (3)	C5—C4—C9—C3	−178.91 (17)
C8—N1—C1—Cl1	−179.53 (12)	C14—C13—C15—C16	0.7 (3)
C3—C2—C1—N1	−0.2 (3)	C12—C13—C15—C16	179.68 (17)
C10—C2—C1—N1	−178.90 (17)	C15—C13—C12—O2	1.3 (3)
C3—C2—C1—Cl1	179.28 (12)	C14—C13—C12—O2	−179.6 (2)
C10—C2—C1—Cl1	0.5 (2)	C15—C13—C12—C11	−179.36 (18)
C15—C13—C14—C18	−0.5 (3)	C14—C13—C12—C11	−0.28 (18)
C12—C13—C14—C18	−179.71 (16)	N1—C8—C7—C6	178.92 (17)
C15—C13—C14—N2	179.64 (15)	C9—C8—C7—C6	−0.5 (3)
C12—C13—C14—N2	0.45 (19)	C9—C4—C5—C6	−0.4 (3)
C11—N2—C14—C18	179.71 (17)	C14—N2—C11—O1	179.81 (17)
C10—N2—C14—C18	−1.1 (3)	C10—N2—C11—O1	0.6 (3)
C11—N2—C14—C13	−0.46 (19)	C14—N2—C11—C12	0.25 (18)
C10—N2—C14—C13	178.72 (15)	C10—N2—C11—C12	−178.93 (15)
C1—N1—C8—C9	0.2 (3)	O2—C12—C11—O1	−0.2 (3)
C1—N1—C8—C7	−179.23 (16)	C13—C12—C11—O1	−179.55 (17)
C11—N2—C10—C2	−98.40 (18)	O2—C12—C11—N2	179.41 (17)
C14—N2—C10—C2	82.5 (2)	C13—C12—C11—N2	0.02 (18)
C3—C2—C10—N2	2.8 (2)	C8—C7—C6—C5	0.7 (3)
C1—C2—C10—N2	−178.58 (16)	C4—C5—C6—C7	−0.2 (3)
N1—C8—C9—C3	0.0 (3)	C13—C14—C18—C17	0.2 (3)

C7—C8—C9—C3	179.40 (16)	N2—C14—C18—C17	180.00 (16)
N1—C8—C9—C4	−179.55 (15)	C13—C15—C16—C17	−0.6 (3)
C7—C8—C9—C4	−0.2 (3)	C15—C16—C17—C18	0.3 (3)
C2—C3—C9—C8	−0.3 (2)	C14—C18—C17—C16	−0.1 (3)

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C3—H3···N2	0.93	2.49	2.842 (2)	102
C17—H17···O2^{i}	0.93	2.48	3.367 (3)	160

Symmetry codes: (i) *x*, −*y*+3/2, *z*+1/2.

Fig. 1

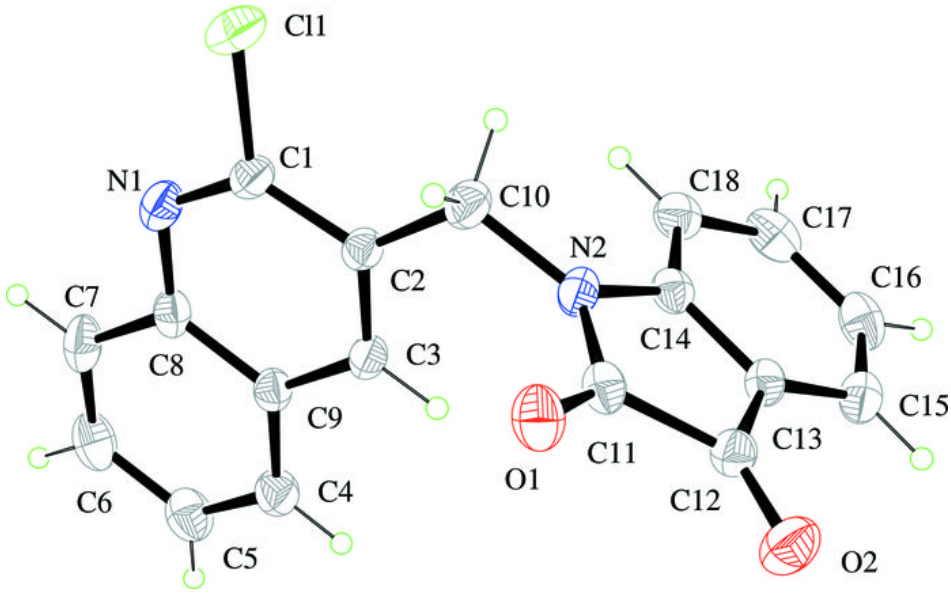
Cl1
C1
N1
C10
C18
C17
N2
C16
C2
C7
C14
C3
C8
C15
C9
C11
C13
O1
C6
C4
C12
C5
O2

Fig. 2